Corrigenda

Fast synthesis of amino acid salts and lactams without solvent under microwave irradiation

Alain Laurent, Patrick Jacquault, Jean-Louis Di Martino and Jack Hamelin

J. Chem. Soc., Chem. Commun., 1995, 1101.

The authors wish to point out that it has been brought to their attention that **6**, when freshly prepared may not be the amino acid hydrogen sulfate but the lactam hydrogen sulfate, which on standing hydrolyses to the amino acid salt. This would account for the presence of a CO_2H group in the IR spectrum and for the formation of lactam **5** on neutralization.

Synthesis and reactivity of novel π-allylcyclopentadienyl cobaltolactone complexes

Steven D. R. Christie, Christophe Cosset, David R. Hamilton, William J. Kerr and John M. O'Callaghan

J. Chem. Soc., Chem. Commun., 1995, 2051.

Para. 3, lines 7–8 on p. 2051 '... the *trans* arrangement of H and H_{endo} results in the ...' should read '... the *trans* arrangement of H-3 and H-4_{endo} results in the ...'

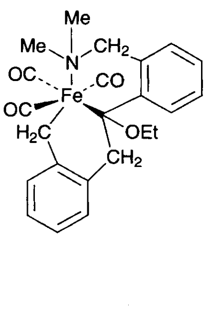
Scheme 2 *Reagents and conditions* on p. 2051 '... v, CAN, CH_2Cl_2, room temp., ...' should read '... v, CTAN, CH_2Cl_2, room temp., ...'

Unusual reaction of tricarbonyl(*o*-quinodimethane)iron with *o*-Lithiobenzyldimethylamine: crystal structure of $[Fe\{Me_2NCH_2C_6H_4C(OEt)CH_2C_6H_4CH_2\}(CO)_3]$

Yong Yu, Jiabi Chen, Jian Chen and Peiju Zheng

J. Chem Soc., Chem. Commun. 1995, 2089.

Structure **2** should be as follows:



Kinetic and stereoelectronic effects of a fluorine substituent on the reaction catalysed by an NADPH-dependent cyclohex-1-enylcarbonyl CoA reductase

Colin F. Bridge,* David O'Hagan,* Kevin A. Reynolds and Kimberlee K. Wallace

J. Chem. Soc., Chem. Commun., 1995, 2329.

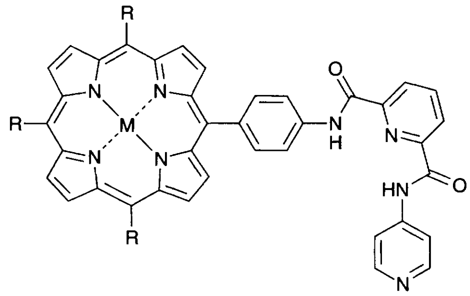
Please note that Kevin Reynolds and *not* Colin Bridge should have been one of the asterisked corresponding authors in this collaborative study between the University of Durham, UK, and the University of Maryland, Baltimore, USA.

Self-assembly of macrocyclic porphyrin oligomers

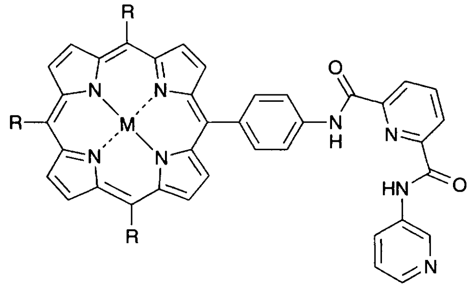
Xianglan Chi, Andrea J. Guerin, Richard A. Haycock, Christopher A. Hunter and Luke D. Sarson

J. Chem. Soc., Chem. Commun., 1995, 2567.

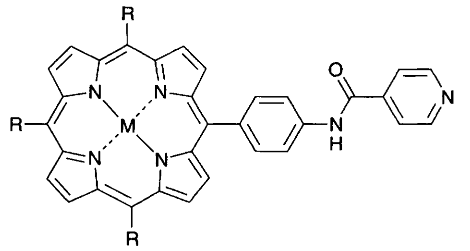
The structures **1a–3b** should be as follows:



1a R = R^1

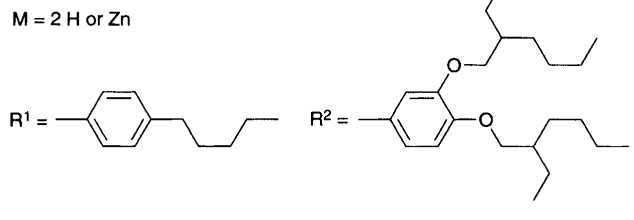


2a R = R^1



3a R = R^1
3b R = R^2

M = 2 H or Zn



Synthesis and surfactant properties of novel fluoroalkylated amphiphilic oligomers

Hideo Sawada, Tokuzo Kawase, Yuka Ikematsu, Yoshinori Ishii, Masatoshi Oue and Yoshio Hayakawa

Chem. Commun., 1996, 179.

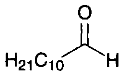
The authors would like to point out that in Fig. 1 and Fig. 2 'Concentration (g dm^{-3})' should read 'Concentration (g/100 ml)'.

Selective cleavage of ketals and acetals under neutral, anhydrous conditions using triphenylphosphine and carbon tetrabromide

Craig Johnstone, William J. Kerr and James S. Scott

Chem. Commun., 1996, 341.

At entry 8 of Table 1 the product should be



The mechanism for the rearrangement of the adamantyl cation

David R. Adams, Patrick D. Bailey, Ian D. Collier, Stephen A. H. Leah and Colin Ridyard

Chem. Commun., 1996, 333.

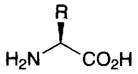
Please note that the fourth author's correct name should be 'Stephen A. J. Leah'.

An efficient route to the α-methyl ester of L-glutamic acid, and its conversion into *cis*-5-hydroxy-L-pipecolic acid

David R. Adams, Patrick Bailey, Ian Collier, John D. Heffernan and Stephen Stokes.

Chem. Commun., 1996, 349.

Structure **3** contains an R group:



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Also RSC Production Division apologies for the omission of David R. Adams from the graphical abstract and author index entries.

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